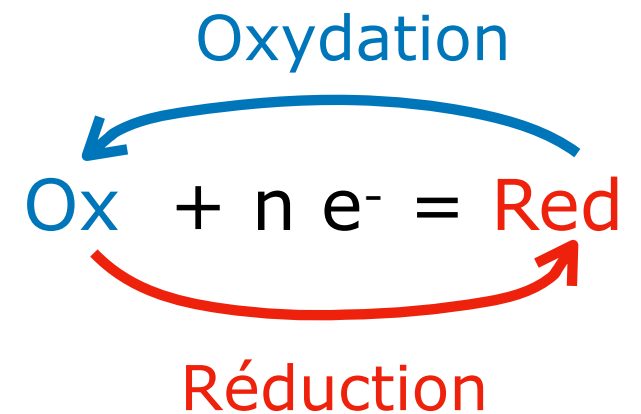
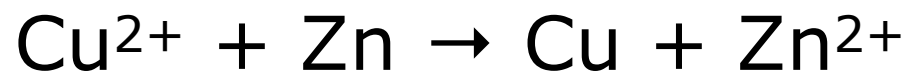
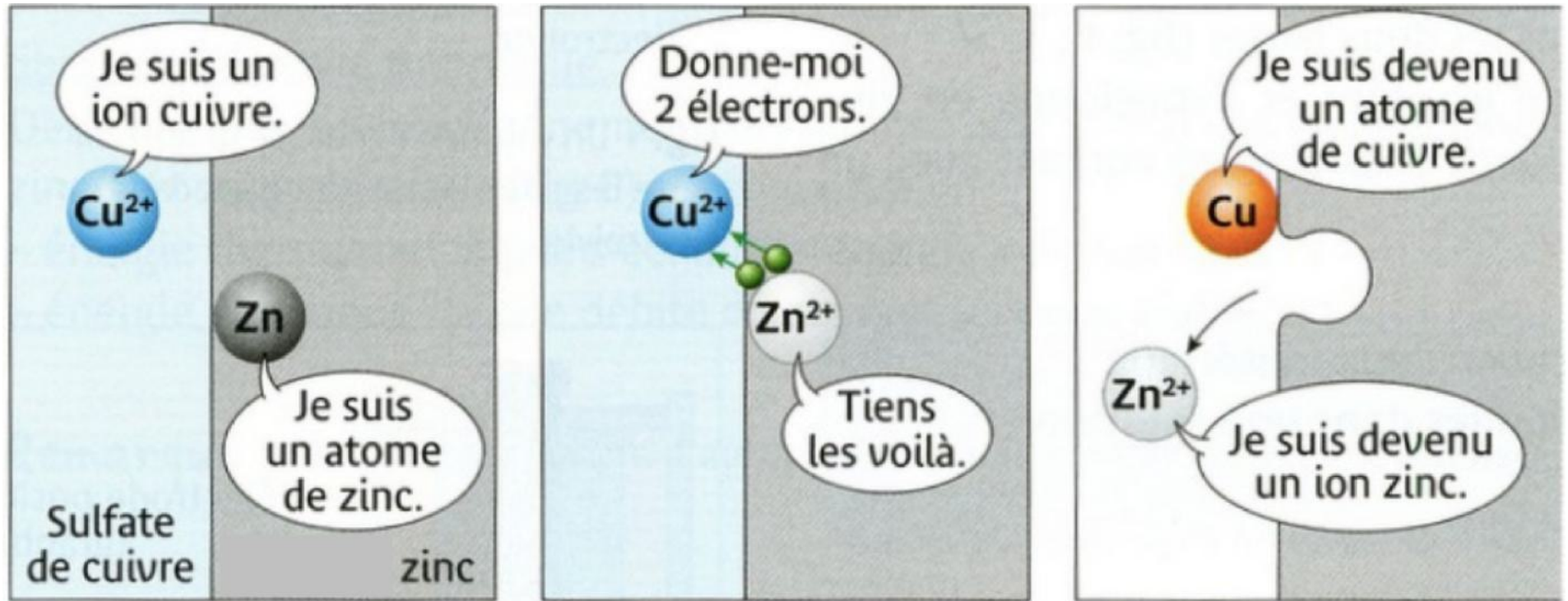
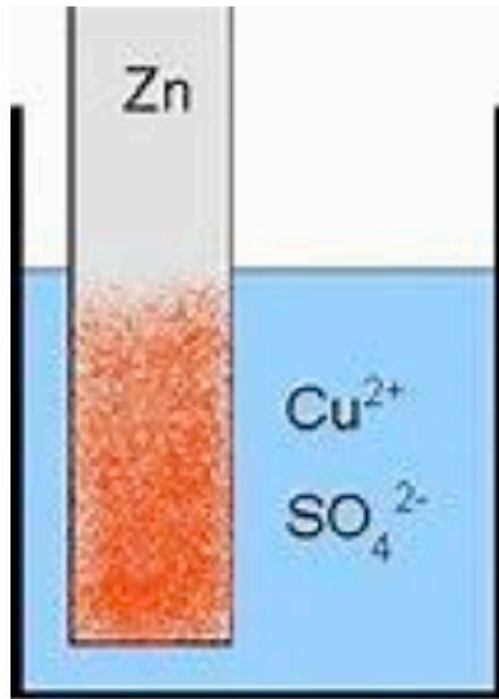
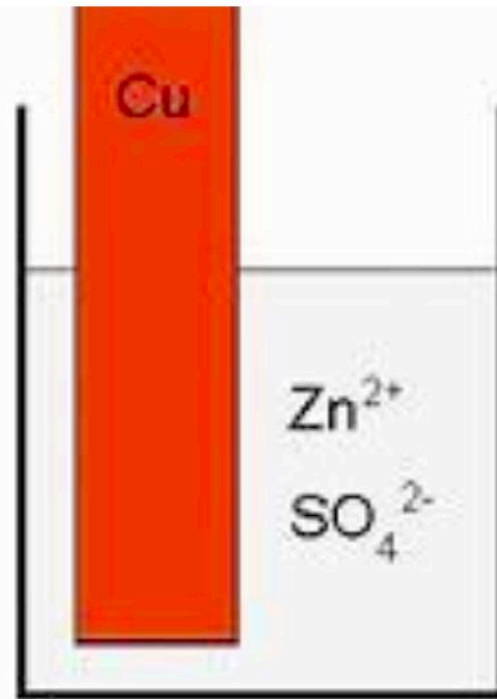


# Oxydation de Zinc par le Cuivre

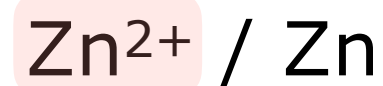
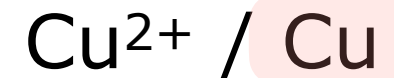
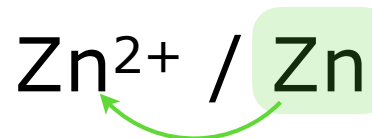
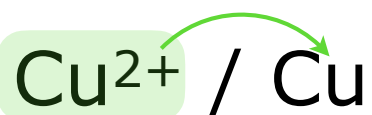




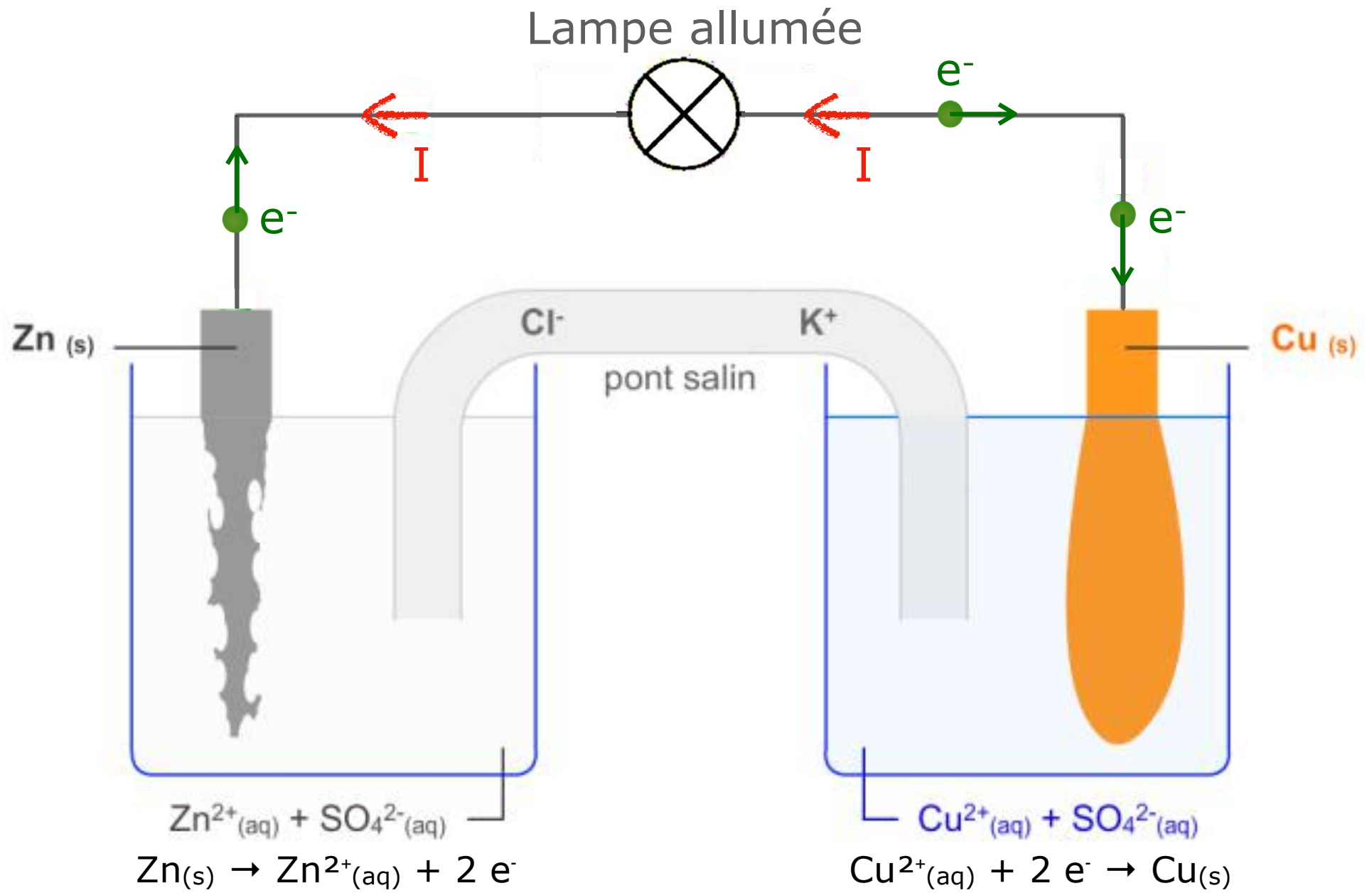
formation d'un  
dépôt de cuivre



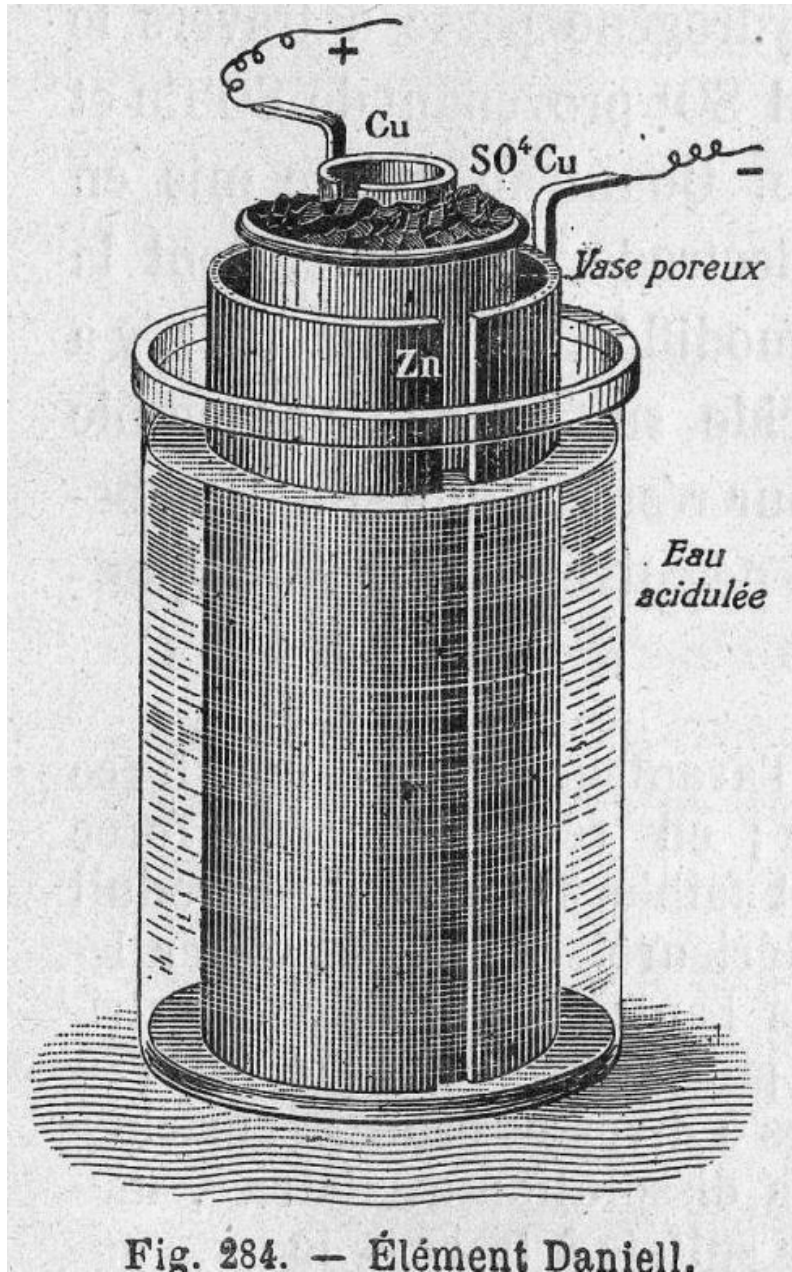
rien



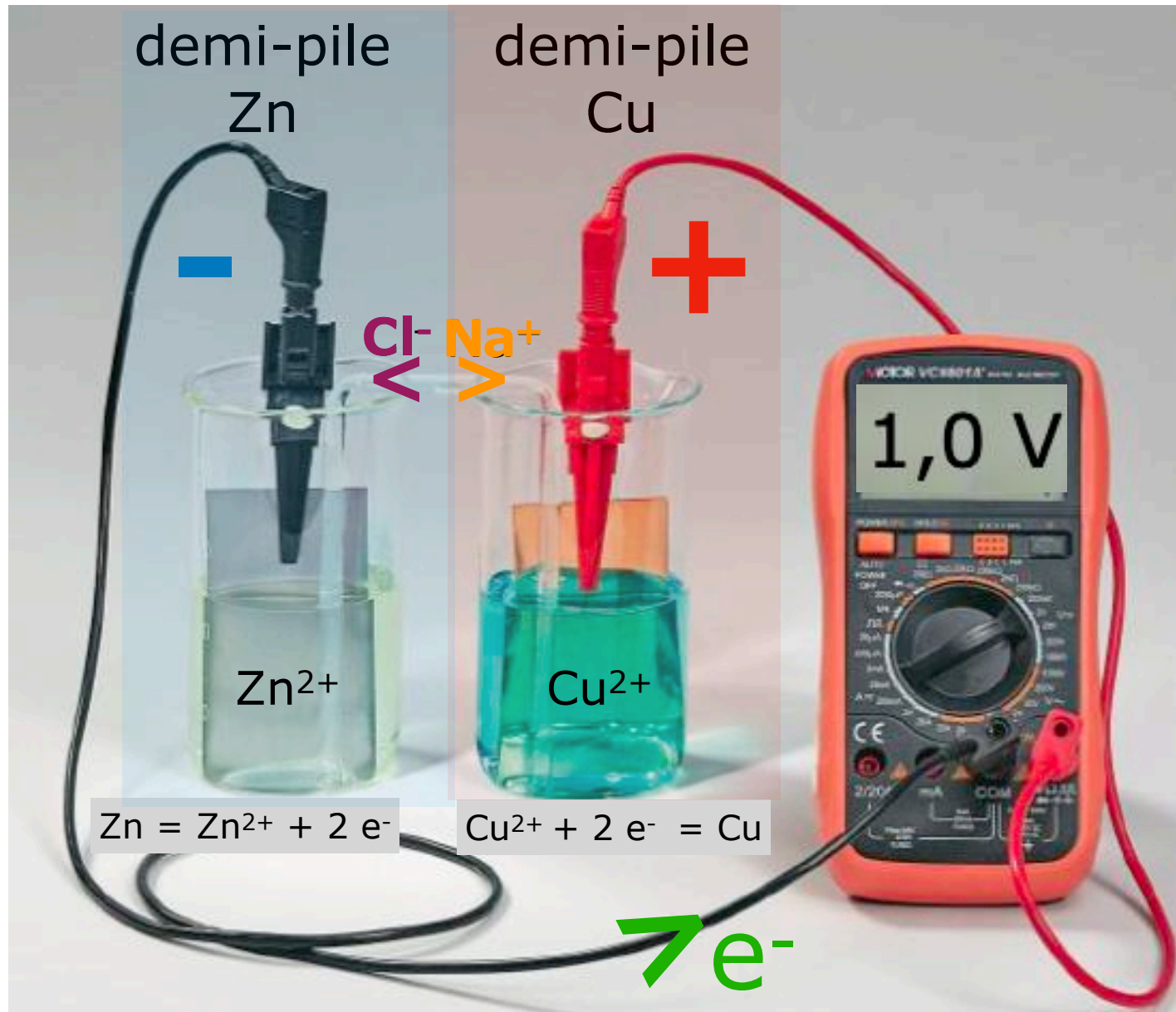
$\text{Cu}^{2+} + \text{Zn} \rightarrow \text{Cu} + \text{Zn}^{2+}$  est le sens spontané de la réaction.  
Dans le sens inverse, la réaction ne se fait pas.







# Pile Daniell



Le pont salin sert à assurer la neutralité électrique des solutions et à fermer le circuit électrique.